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The Syntheses and Binding Properties of the Novel Organophosphorus Calixarenes

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Phosphorylation of bromo, dialkylaminomethyl calix[4]resorcinarenes with various long chain aliphatic radicals with a number of P(III) derivatives i.e. triamido- and diamidophosphites, and polyhalogenides of P(III), P(IV) is described. A series of phosphoamide cavitands were obtained. Structures and properties of synthesized compounds were discussed on the basis of physical and quantum chemical methods.

Keywords: Phosphorylation; calix[4]resorcinarenes; carceplex; cavitands

INTRODUCTION

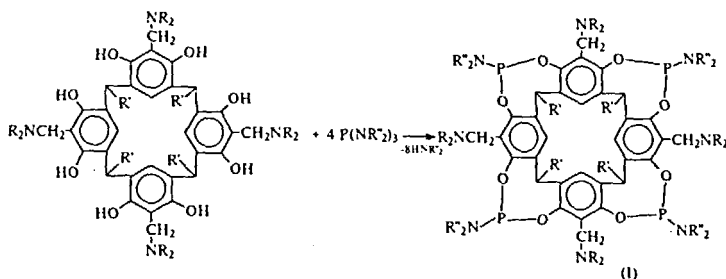
Cyclodextrins, crown ethers and calixarenes are prominent in host-guest chemistry. Calixarenes and their derivatives can selectively involve both organic and inorganic guests. Bromo, dialkylaminomethyl calix[4]resorcinarenes are of the great interest because of their accessible synthesis, sufficient solubility and "bowl-like" conformation. The molecular design of new three-dimensional structures with calixarene as a basis can be achieved with the use of a wide range of reagents and methods of organophosphorus chemistry [1]. The purpose of this work is synthesis and investigation of new three-dimensional structures with three- and four-coordinated phosphorus which were obtained on the basis of bromo, dialkylaminomethyl calix[4]resorcinarenes.

RESULTS

The phosphorylation of dialkylaminomethylated calix[4]resorcinarenes with P(III)-amides results in the formation of structures with cyclic phosphorus fragments fixed on the cavity rim [2,3], which are easily hydrolyzed.

It was found that hydrophosphoryl derivatives of the dialkylaminomethyl-

calix[4]resorcinarene undergo two types of transformations depending on the type of substituent in dialkylamino groups: heterocyclization with elimination of 4 molecules of water and formation of phosphor(III)ylated cavitands or heterocyclization with elimination of 4 molecules of dialkylamine and formation cavitands with cyclic hydrophosphoryl groups [4].



R=CH₃, C₂H₅, C₃H₇.

R'=CH₃, C₂H₅, C₃H₇, C₆H₁₃, C₁₁H₂₃.

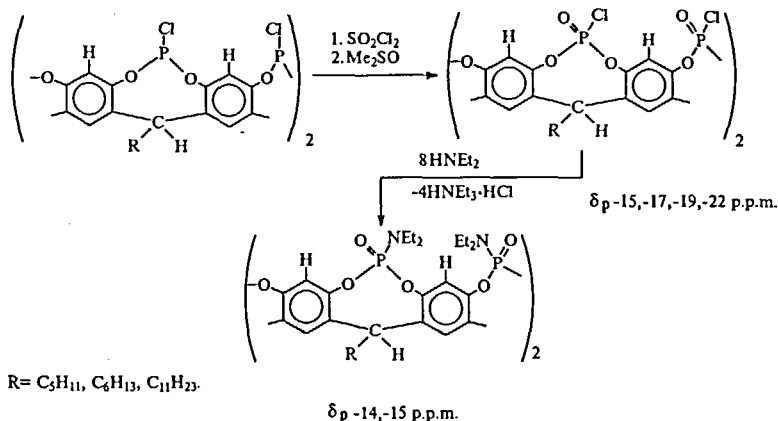
R''=CH₃, C₂H₅.

δp 139-143 p.p.m.

The interaction of dialkylaminomethylated calix[4]resorcinarene with tetraalkyldiamidoalkylphosphites proceeds with formation of sault type product (initial calixarene with alkylphosphoric acid).

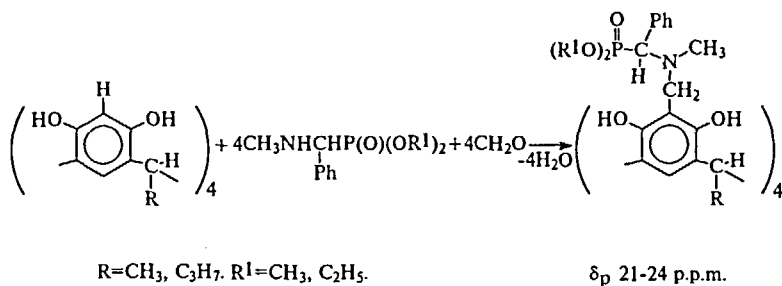
The interaction of calix[4]resorcinarenes, bearing aliphatic radicals of different length, with phosphorus trichloride proceeds in different manner, depending on reaction conditions. When 1:8 (calixarene : PCl₃) ratio have been used the cyclic hydrophosphoryl cavitands have been obtained. In the presence of a base (1:8:3 calixarene : PCl₃ : triethylamine) the final products are cyclic chlorophosphites. They don't react with trichlorophosphorus thiooxide, but may be easily oxidized into P(IV)-derivatives by sulfurylchloride or dimethylsulphoxide.

As a result of quantum calculation, it was shown that length of P - Cl bond in cyclic chloro P(III) derivatives of calix[4]resorcinarenes is much more shorter as compare with chloro P(III) phosphosines. This fact explains the anomal chemical shift of cyclic chloro P(III) derivatives of calix[4]resorcinarenes in NMR ³¹P spectra (127 - 129 p.p.m.).

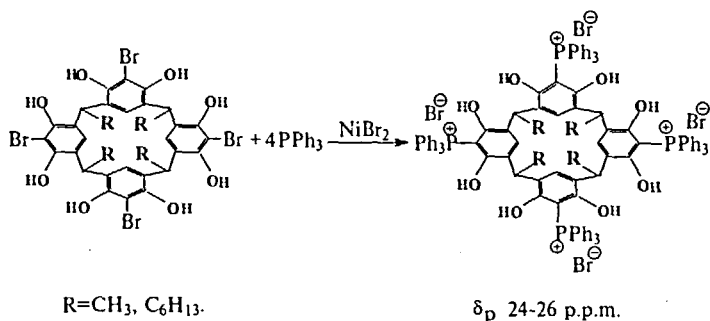


Unlike phosphorus trichloride the phosphorylation of calixarenes with trichlorophosphorus oxide proceeds with the formation only linear compounds. Depending on the calixarene - trichlorophosphorus oxide ratio two or four dichlorophosphate groups can be introduced into the molecule of calixarene. When heated they are not converted into cyclic derivatives. According NMR ^{31}P spectra the complicated mixture of different phosphorus containing compounds have been formed.

New C-phosphorylated alkylarylaminoethyl cavitands have been obtained as a result of the reaction of calix[4]resorcinarenes, α -aminophosphonates, formaldehyde in the 1:4:4 ratio [5].



The new type phosphonium salts has been obtained as a result of phosphorylation of brom derivatives of the calix[4]resorcinarenes with triphenylphosphine in the presence of catalytic amounts of NiBr_2 [6].



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